A SIMPLE WAY OF CALCULATION OF SPIN DENSITIES FROM RHF WAVE FUNCTION

Jozef TIŇO^a, Viliam KLIMO^a and Michael M. MESTECHKIN^b

^a Institute of Polymers, Slovak Academy of Sciences, 809 34 Bratislava and ^b Institute of Physical Organic Chemistry, Ukrainian Academy of Sciences, 340 048 Donetsk, USSR

Received December 22nd, 1979

From study of non-stability of the RHF method a formula has been derived for addition of the spin-polarization component to the RHF spin densities. Spin densities of several radicals have been calculated by the INDO method. The results are compared with those of RHF CI and PUHF, the agreement being very good.

To obtain even qualitatively correct spin densities, the RHF method must be extended by configuration interaction or by perturbation method. Increasing number of electrons results in the problem of limitation of number of the used configurations and restriction of order of the perturbation method in the former and the latter method, respectively. Hence, although theoretical state of calculation of spin densities is clear in principle at present, it continues to be a topical problem to find a simple way giving not only a correct qualitative picture of spin distribution in a system but also the best possible quantitative results. Such proposal was recently submitted by Amos and coworkers¹ with the aim to improve the spin densities obtained from the UHF wave function. The aim of this communication is to suggest a simple method of correction of the 1. order to the spin densities obtained by the RHF method.

THEORETICAL

Studying non-stability of the RHF method, one of the authors (M. M. M.) arrived at the conclusion that the spin density matrix plays an important role in evaluation of external non-stability in the RHF method². For introduction of the used symbolism and for better understanding of the derived relation we give a brief account of the basic ideas leading to the derived formalism.

It was shown³ that the generalized version of HF method can be reduced to calculation of two matrixes Y and Z which represent (in restricted and unrestricted cases) matrices of residual charge densities and spin densities, respectively, and fulfil the following conditions:

$$[\mathbf{Y}, \mathbf{Z}]_{+} = 0, \quad \mathbf{Y}^{2} + \mathbf{Z}^{2} = \mathbf{I}, \quad Sp\mathbf{Z} = 2M, \quad Sp\mathbf{Y} = N - m, \quad (1)$$

where N, M, and m are number of electrons, spin projection value, and dimension of basic set, respectively. Variation of energy to the 1. order and conditions of the extreme are as it follows³:

$$\delta E = Sp(\mathbf{F}_1 \delta \mathbf{Y} + \mathbf{F}_2 \delta \mathbf{Z}) \tag{2}$$

$$[F_1, Y]_- + [F_2, Z]_- = 0, [F_1, Z]_- + [F_2, Y]_- = 0$$
(3)

The ref.³ gives discussion about the form of the operators F_1 and F_2 which are functions of **Y** and **Z** for the restricted and unrestricted HF method. In the restricted HF method a further condition must be fulfilled:

 $Z^2 = Z$

and hence $\mathbf{Z} = \mathbf{I} - \mathbf{Y}^2$ and $\mathbf{Y}^3 = \mathbf{Y}$.

TABLE I

Spin Densities of Some Radicals Calculated by INDO Method in Two Parametrizations by RHF CI Method, and according to Eq. (9)

Padical	Atom	Pople ⁵		Kaufman ⁶	
Radical	Atom –	(9)	RTH CI	(9)	RTH CI
CH ₃	н	0.0215	0.0278	0.0326	-0.0414
C ₂ H ₅	$^{H_1}_{H_2}$	0·0193 0·0349	0·0256 0·0463	0·0293 0·0341	0·0379 0·0475
НСО	н	0.1485	0.1990	0.1390	0.1817
HC=CH ₂	H ₁ H ₂ cis H ₂ trans	0·0325 0·0769 0·0313	0·0347 0·1304 0·0443	0·0226 0·0733 0·0279	0·0190 0·1269 0·0405
NH ₂	н	0.0206	0.0241		0.0363
$\overset{H_1}{\underset{H_3}{\overset{\downarrow}{}}} C^{\mathcal{H}_{\mathcal{L}}_{\mathcal{L}}_{\mathcal{L}_{\mathcal{L}_{\mathcal{L}}_{\mathcal{L}_{\mathcal{L}}_{\mathcal{L}_{\mathcal{L}}_{\mathcal{L}_{\mathcal{L}}_{\mathcal{L}_{\mathcal{L}}_{\mathcal{L}_{\mathcal{L}}_{\mathcal{L}}_{\mathcal{L}}_{\mathcal{L}}}}}}}}}}$	H1 H ₂ H ₃	0.00002 0.0103 0.0104	0.0003 0.0130 0.0135	0·00004 0·0157 0·0159	0.0006 0.0197 0.0208
$CH_3NO_2^-$	Н	0.0125	0.0135	0.0127	0.0135
NH ₂ —NH ₂ ⁺	Н	0.0133	0.0171	0·0189	0.0245
$\underset{CH_{3}}{\overset{H_{2}}{\succ}}N-N\overset{H_{3}^{+}}{\overset{H_{4}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}$	H ₁ (CH ₃) H ₂ H ₃ H ₄	0.0140 0.0167 0.0082 0.0083	0.0149 0.0163 0.0087 0.0099	0.0130 0.0227 0.0129 0.0130	0.0131 0.0228 0.0137 0.0156

For calculation of the F_1 and F_2 operators we must know the matrix of the oneelectron part of hamiltonian, the matrices K and G defined in ref.³

$$K_{pq}(\mathbf{Z}) = -\frac{1}{2} \sum_{t,r} (pr \mid tq) Z_{tr}$$
(4)

$$G_{pq}(\boldsymbol{Z}) = \sum_{t,t} \left[\left(pr \mid qt \right) - \frac{1}{2} \left(pr \mid tq \right) \right] Z_{tr}$$
(5)

and matrices Y and Z which, in the UHF and RHF schemes, are calculated according to Eq. (6)

$$Y = C_{\alpha}C_{\alpha}^{+} + C_{\beta}C_{\beta}^{+} - I$$
$$Z = C_{\alpha}C_{\alpha}^{+} - C_{\beta}C_{\beta}^{+}, \qquad (6)$$

where C_{α} and C_{β} are matrices of coefficients for the orbitals occupied by electrons with α and β spins, respectively. On the basis of the study of non-stability of the RHF method in ref.² the relation (7) was obtained for the energy variation to the first order for the RHF solution:

$$\delta E_0 = \frac{1}{2} S p \, \mathbf{Y} [\mathbf{Y} \mathbf{K} (\mathbf{Z}) \, \mathbf{Y} - \mathbf{K} (\mathbf{Z})] \, \mathbf{Y} \delta \mathbf{Z} \,, \tag{7}$$

where $\delta \mathbf{Z}$ can be considered independent, and thus the choice:

$$\delta \mathbf{Z} = -k\mathbf{Y}[\mathbf{Y}\mathbf{K}(\mathbf{Z})\mathbf{Y} - \mathbf{K}(\mathbf{Z})]\mathbf{Y}, \quad k > 0$$
(8)

ensures the inequality $\delta E_0 < 0$.

TABLE II

Analysis of Linear Dependence between Experimental *hfs* Constants and Spin Densities Calculated in INDO Approximation by RHF CI Method, PUHF Method and according to Eq. (9) in Two Parametrizations

Method	Number of points	Correlation coefficient	Standard deviation, mT	K _N mT
RHE CI ⁴	17	0.991	0.55	65.51
$(9)^b$	17	0.993	0.24	92.37
$(9)^{a}$	17	0.996	0.42	86.21
PUHE	17	0.996	0.42	

Parametrization: " Kaufman⁶, ^b Pople⁵.

Collection Czechoslov, Chem. Commun. [Vol. 45] [1980]

For the complete matrix of spin densities relation (9) is obtained from Eq. (8).

$$\varrho = \mathbf{Z} + \delta \mathbf{Z} \,. \tag{9}$$

Calculation Method

To enable comparison of the spin densities calculated from Eq. (9) with results of the RHF CI method, we chose the same set of radicals as that used for the reference RHF CI calculation⁴ (Table I). The spin densities were calculated by the INDO method in standard parametrization suggested by Kaufman and coworkers⁶.

1

RESULTS AND DISCUSSION

The relation (8) contains the constant k which must be determined additionally. It is the same for all matrices of spin densities and can be determined from the condition for the energy minimum in the given method.* In our case the calculation is a semiempirical one, the constant of proportionality between hfs constants and spin density

$$a_{\rm N} = K_{\rm N} \varrho_{\rm sN} \tag{10}$$

being determined by the least squares method⁵. In our calculations it was presumed k = 1; additional comparison of the RHF CI results with our results shows that $k \sim 1$ (Table I). From Table I it follows that the spin densities calculated from Eq. (9) agree well with the values of the RHF CI method even without optimization of the k constant. As the calculation is semiempirical, Table II gives evaluation of the results by the least squares method. With the both parametrizations used the correlation coefficients are very good, being comparable with the best results of the UHF type calculation⁴ and better than the RHF CI results. With respect to simple calculation this method can serve as a rapid and sufficiently accurate information source about spin distribution in open-shell systems.

REFERENCES

- 1. Amos A. T., Beck D. R., Cooper I. L.: Chem. Phys. Lett. 56, 95 (1978).
- 2. Mestechkin M. M.: Int. J. Quantum Chem. 13, 469 (1978).
- 3. Mestechkin M. M., Whyman G. E.: Int. J. Quantum Chem. 8, 45 (1974).
- 4. Klimo V., Tiňo J.: Chem. Phys. Lett. 37, 311 (1976).
- Pople J. A., Beveridge D. L.: Approximate Molecular Orbital Theory. McGraw-Hill, New York 1970.
- 6. Kaufman J. J., Predney R.: Int. J. Quantum Chem. 6, 231 (1972).

Translated by J. Panchartek.

^{*} The problem of correct choice of the k constant will be discussed in a following *ab initio* calculation of spin densities using this method.